



# An Examination of the Vulnerability of Groundwater to Climate Change in Olympic Valley

**Jean E. Moran (California State University East Bay)**

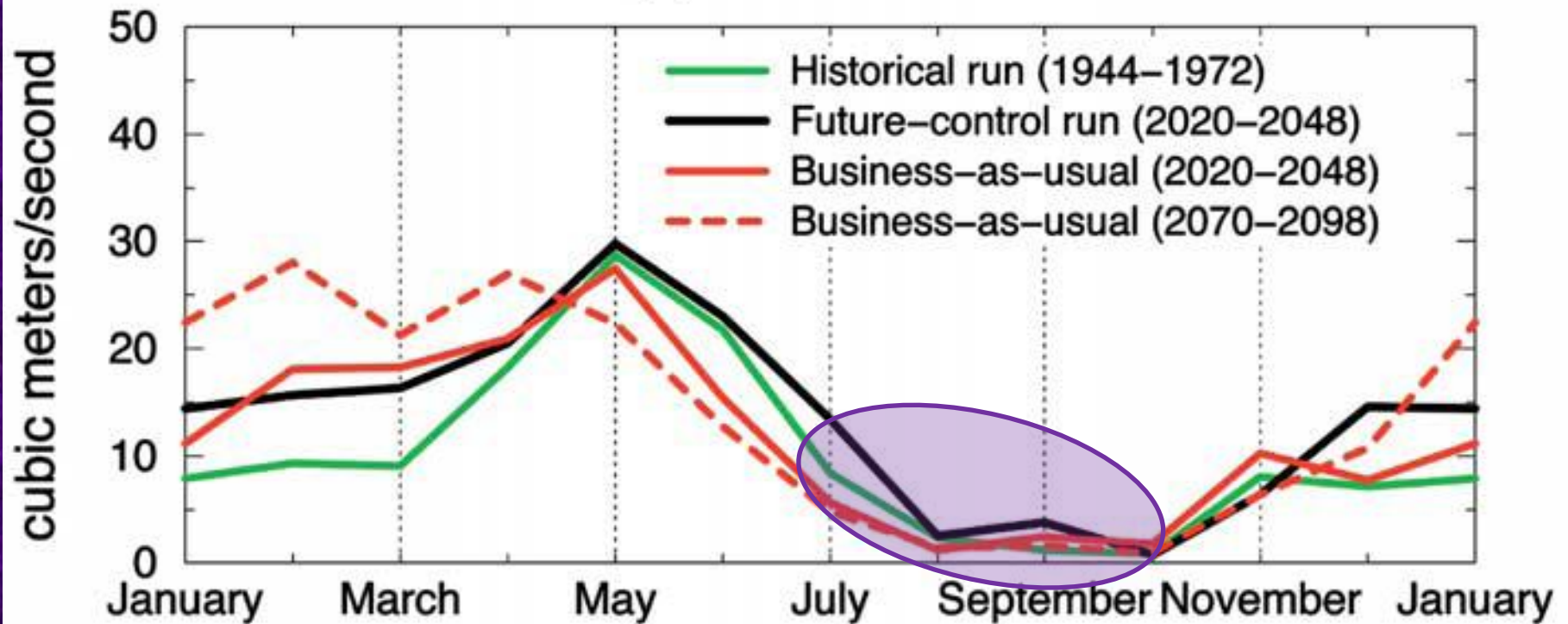
**Michael J. Singleton, Darren Hillegonds, Ate Visser, Brad  
Esser (Lawrence Livermore National Laboratory)**

# Presentation Outline

- Predicted effects of climate change on runoff and groundwater
- Effects particular to alpine/subalpine (snowmelt dominated) groundwater basins
- Dissolved gas and isotopic tools applied
- Results from Olympic Valley groundwater basin
- Implications for recharge under warmer climate conditions

# High Certainty for Earlier Peak Streamflow

MEAN-MONTHLY STREAMFLOW  
(a) Merced River



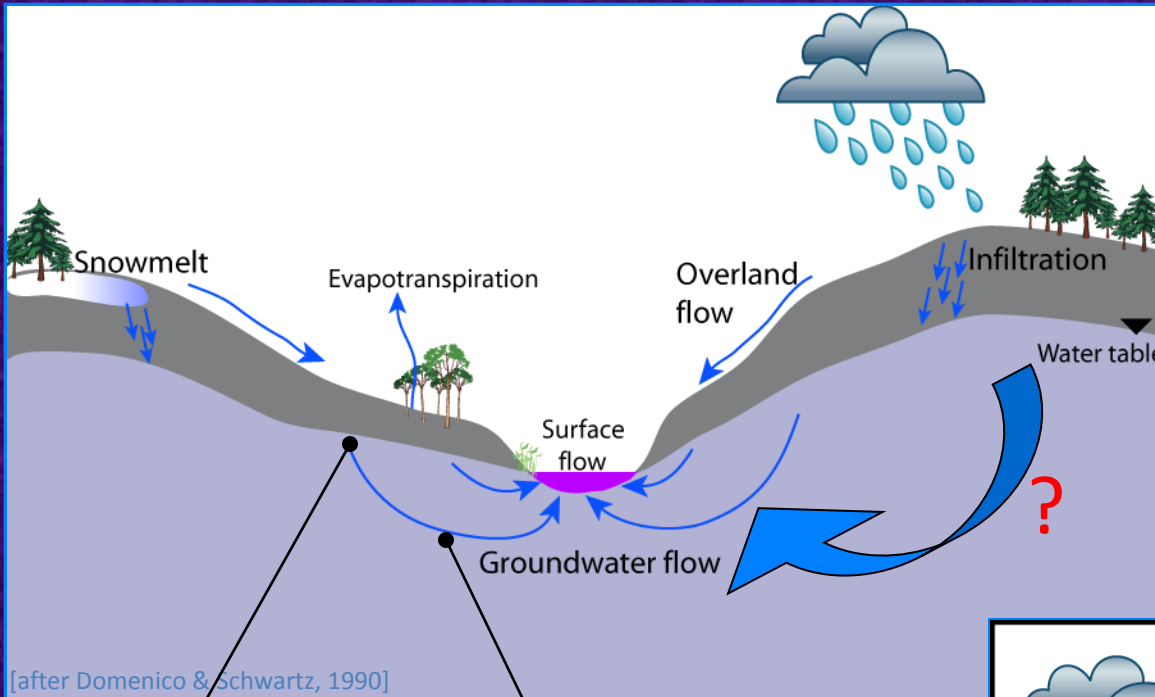
From: Dettinger et al., 2004



# Challenges in predicting effects of climate change on groundwater

- Recharge is strongly influenced by changes in precipitation amount, which is not as well-predicted as temperatures
  - Small changes in precipitation may result in large changes in recharge in semi-arid, arid climates
- Downscaling is major issue for predicting GW response
- Wide range in subsurface residence times of complicates response of surface water-groundwater interaction
- Non-climatic drivers exert large influence on recharge and groundwater levels

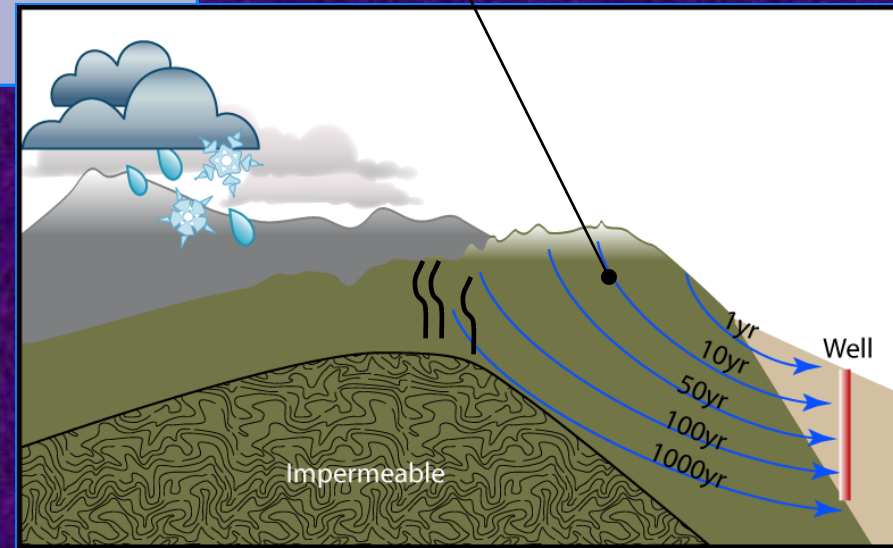
# Connections between snowmelt and groundwater recharge are poorly understood



When and where does recharge take place?  
What is the residence time of groundwater?

**Groundwater Age**  
 $^3\text{H}$ - $^3\text{He}$ ,  $^4\text{He}_{\text{rad}}$

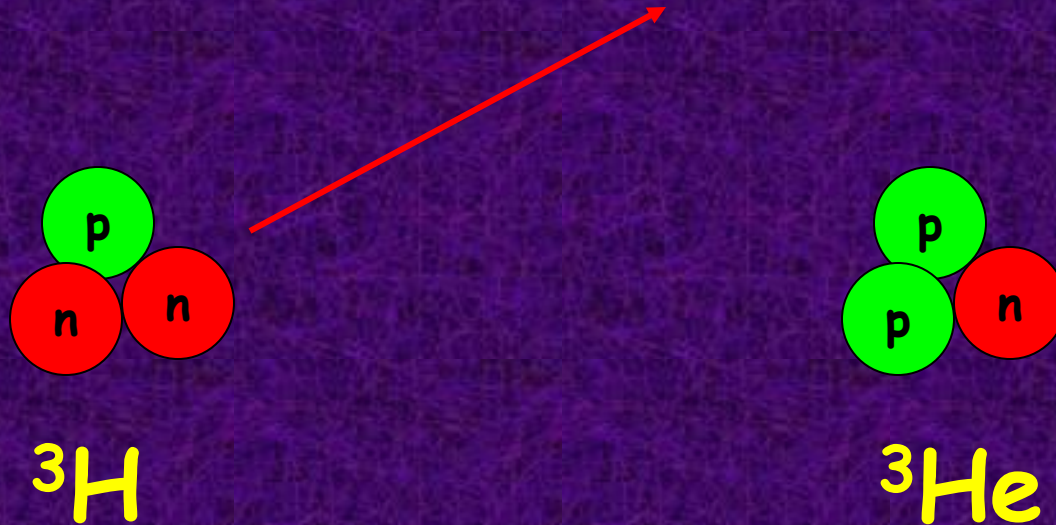
**Noble Gases**  
**Recharge Temperature**  
**Excess Air**



# Climate change effects that are particular to alpine/sub-alpine GW basins

- Occur at elevations where change in form of precipitation will be important
- Rain on snow events generate flooding events
- Down-scaling of GCMs important to capture basin physiography
- Disparate recharge mechanisms possible (mountain block/fractured rock, influent streams, diffuse snowmelt)

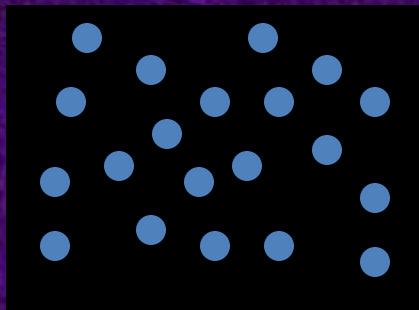
# Tritium decays to $^3\text{He}$



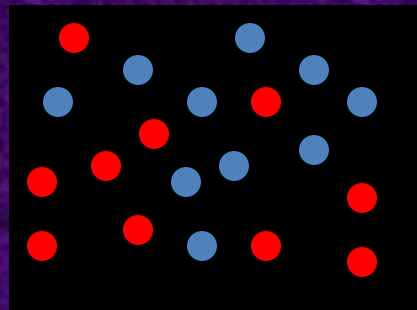
Tritium ( $^3\text{H}$ ) is an unstable nucleus and ejects an energetic electron to become an atom of helium-3 ( $^3\text{He}$ )



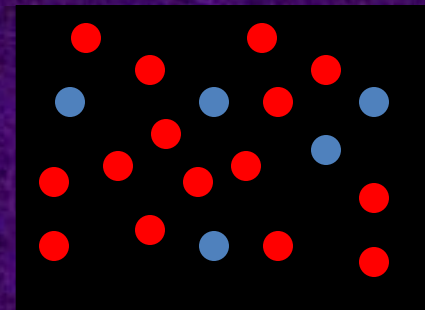
The  $^3\text{He}$  from  $^3\text{H}$  decay starts to accumulate once the water has become groundwater



0 years



12 years



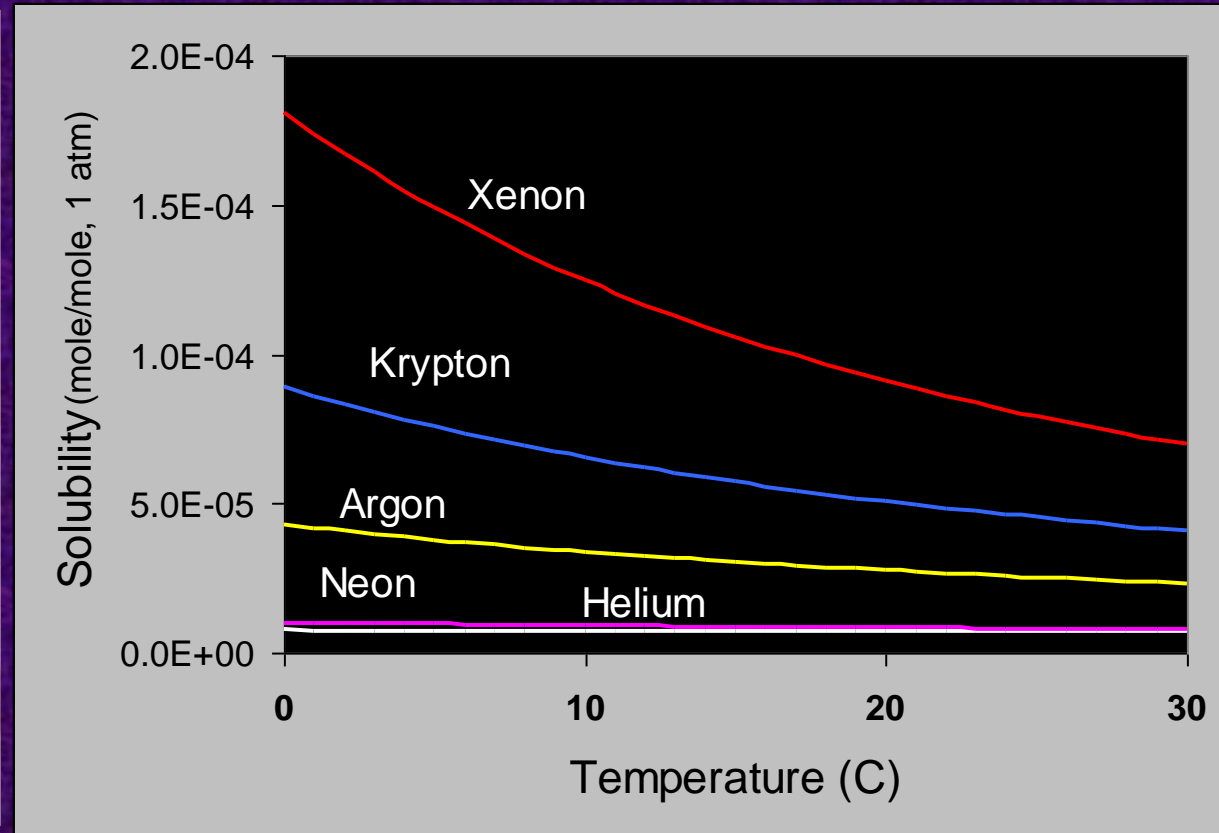
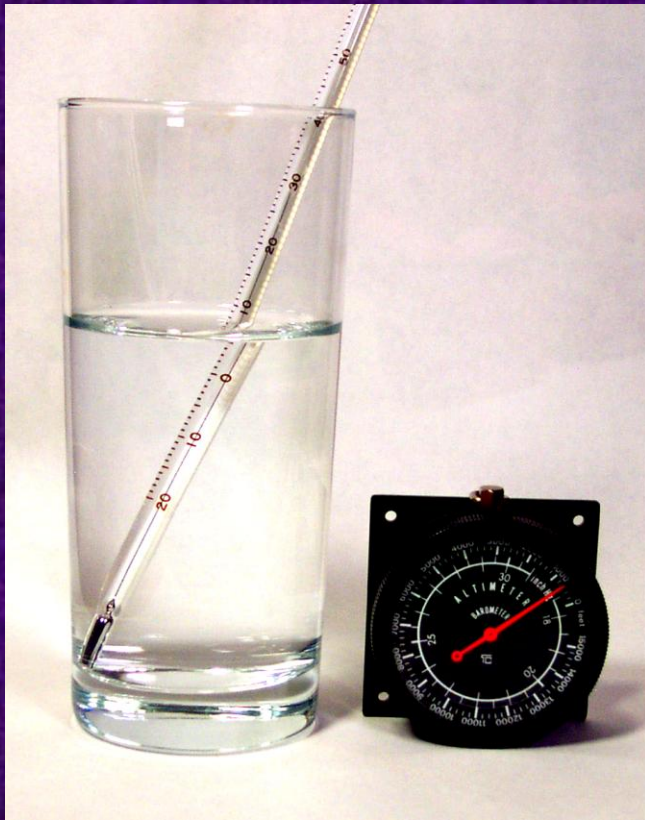
24 years

$$\text{Age (years)} = 18 \times \ln(1 + ^3\text{He} / ^3\text{H})$$



# Temperature of recharge is determined from noble gas concentrations

A reasonable range in pressure (altitude) is assumed and temperatures are calculated from equilibrium solubility component

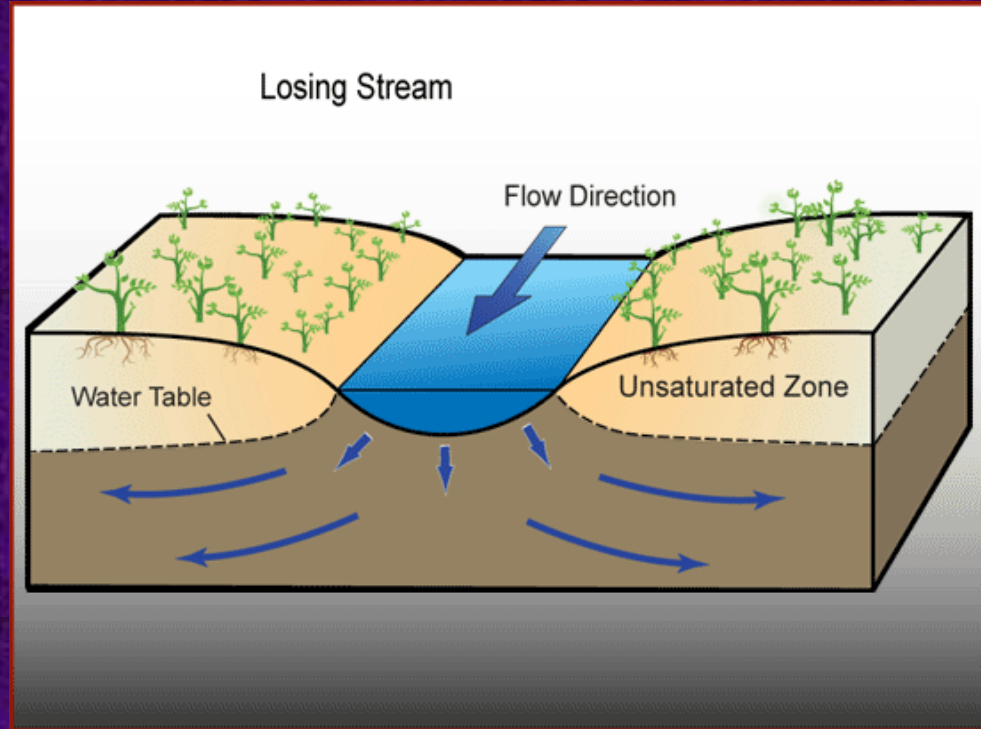


Analytical uncertainty is approximately +/- 1°C

# Excess air concentrations reflect air entrainment and hydrostatic pressure during recharge



**Artificial recharge and recharge through fractures adds a lot of excess air to groundwater due to large fluctuations in the water table**

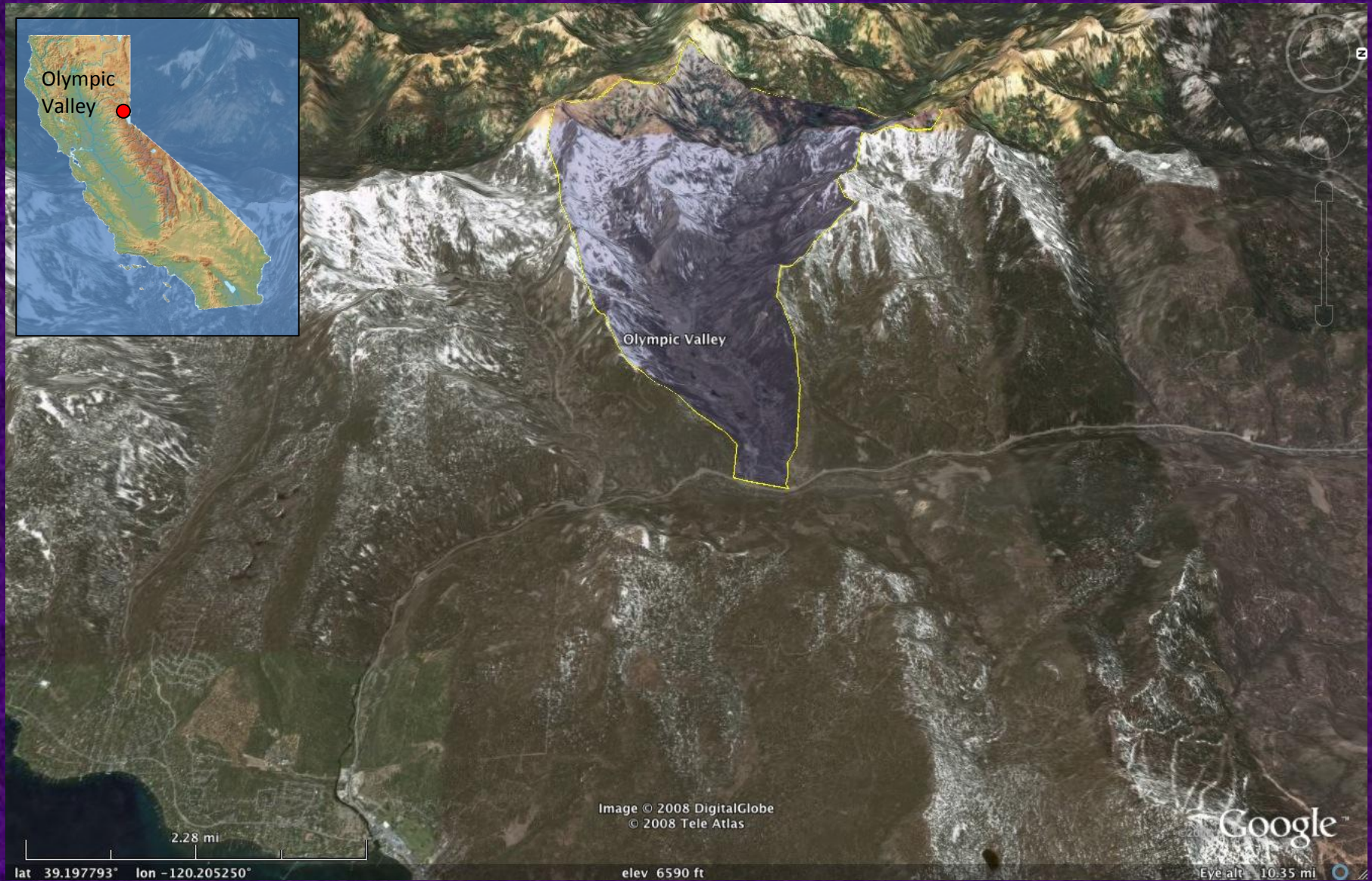


**Little or no vadose zone interaction results in very low excess air**

*Recharge via fractures in hard rock terrain would trap a lot of excess air*



# The Olympic Valley Basin





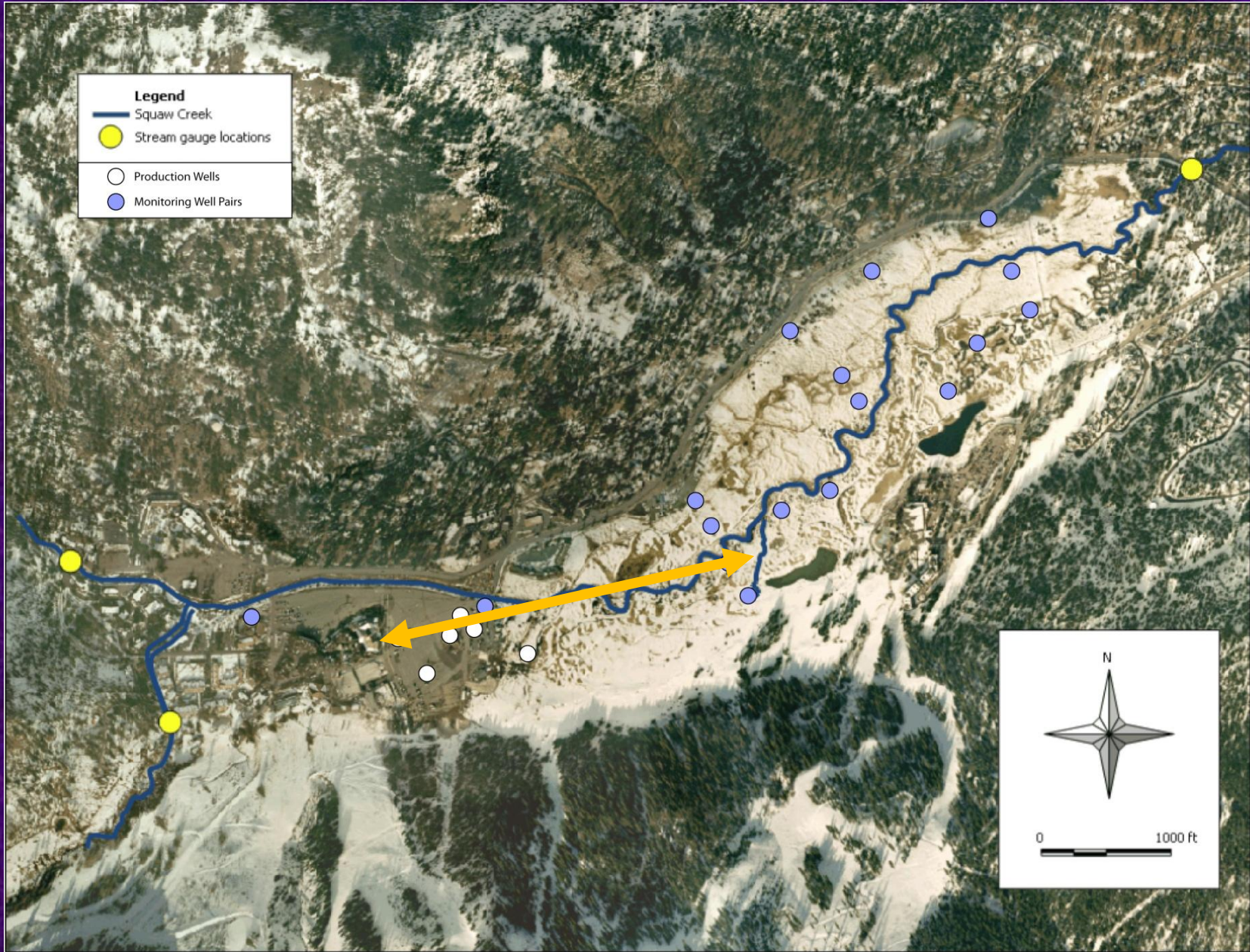
# Wells Sampled

## Olympic Valley

- 6 production wells
- 22 monitoring wells





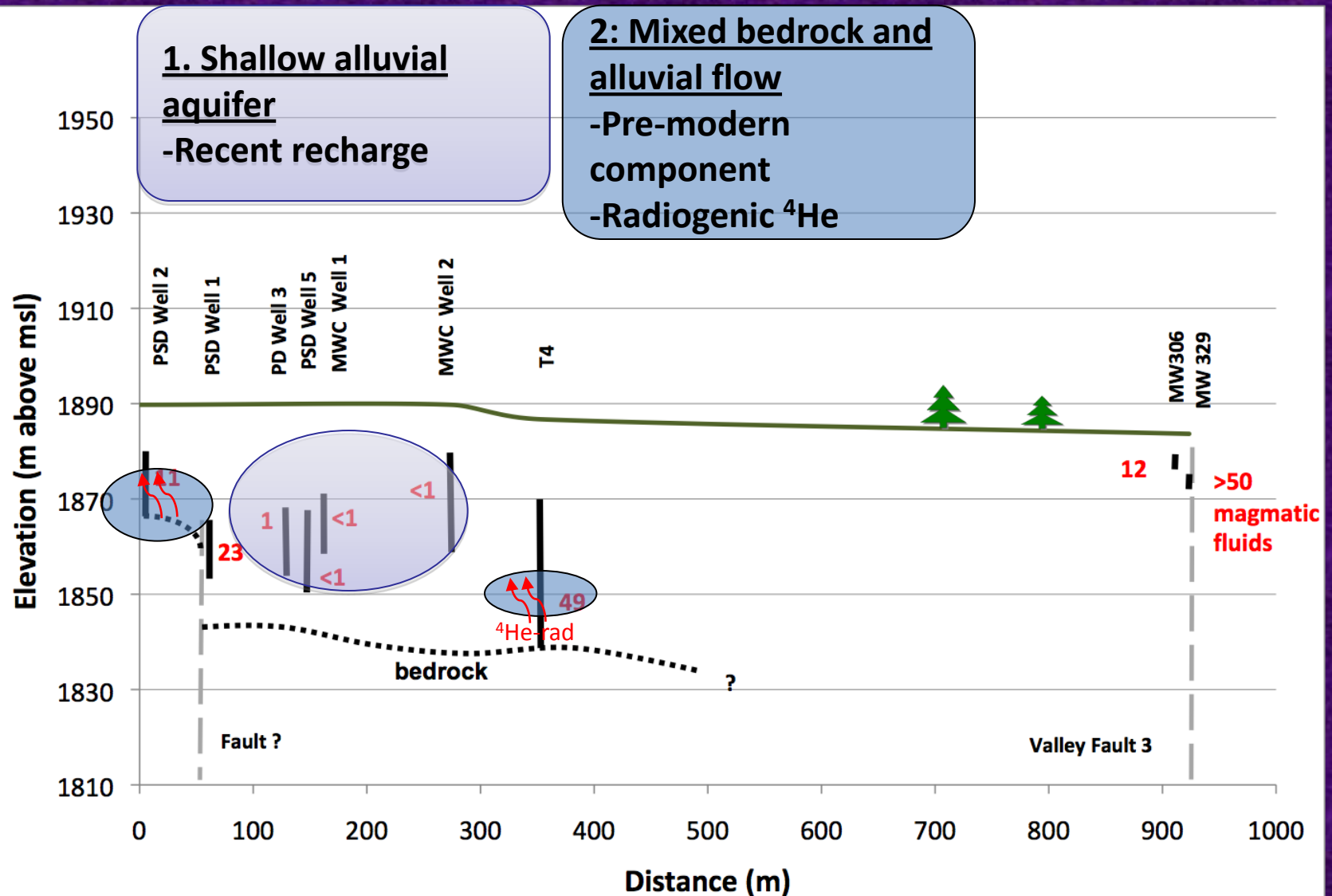


**3 Stream Flow Gauges**

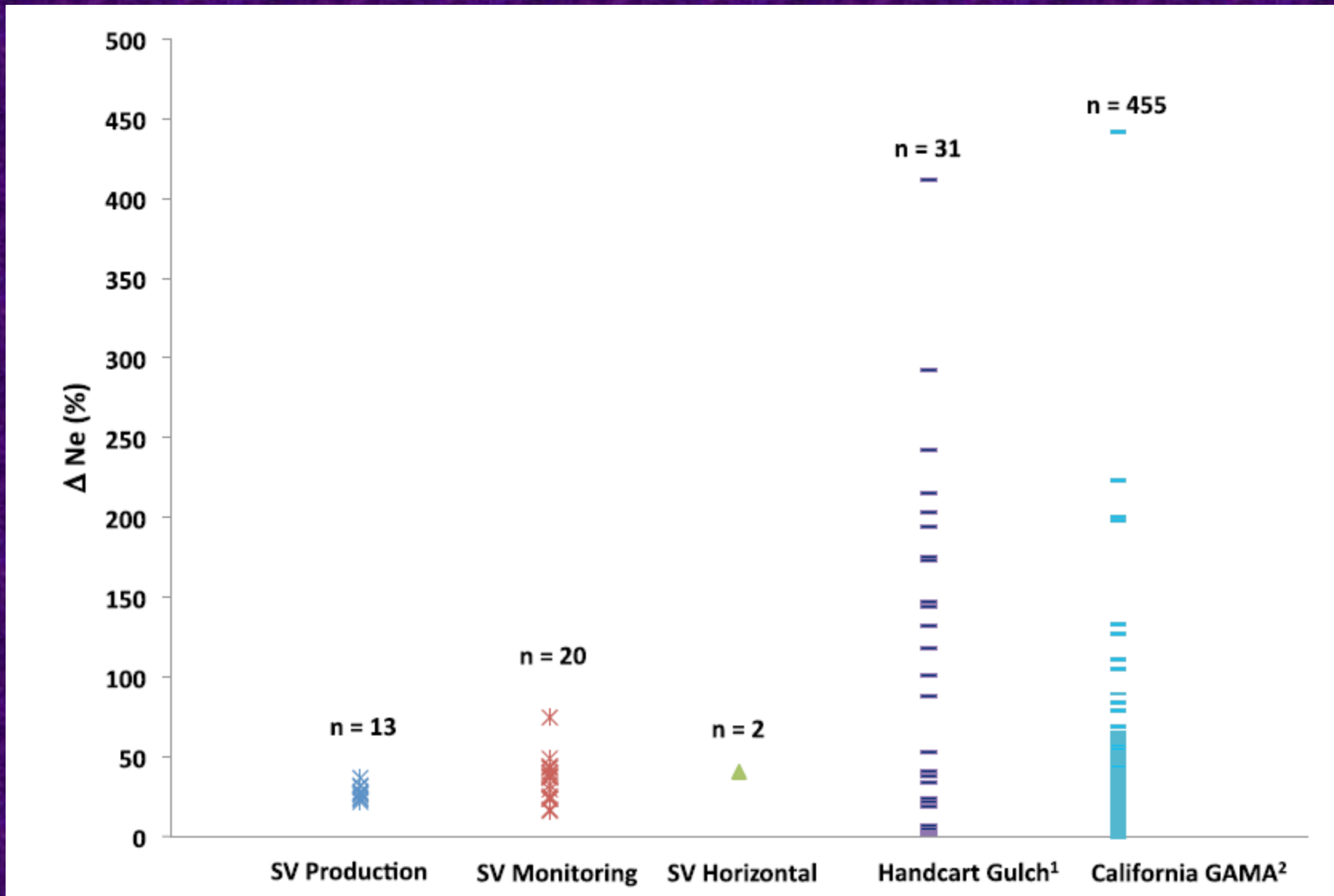
**2 Horizontal Wells**



# Groundwater Ages - Cross Section



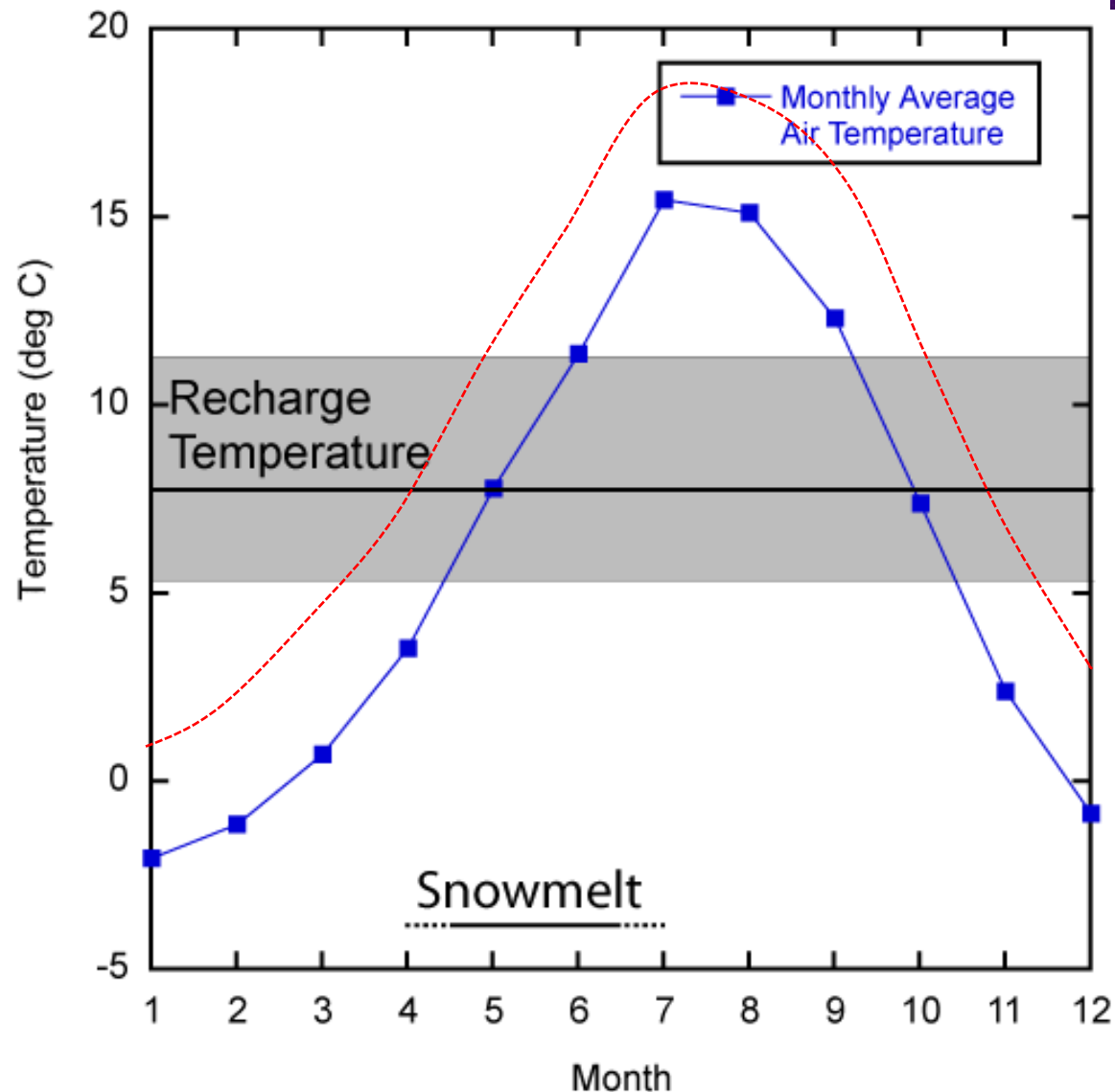
# Comparing excess air concentrations



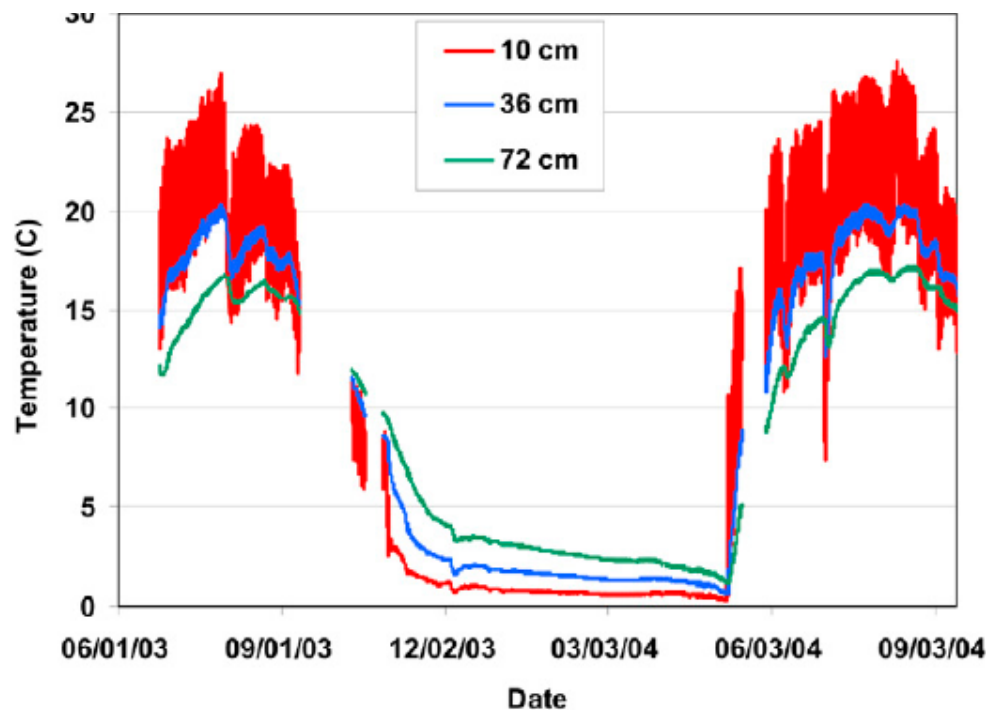
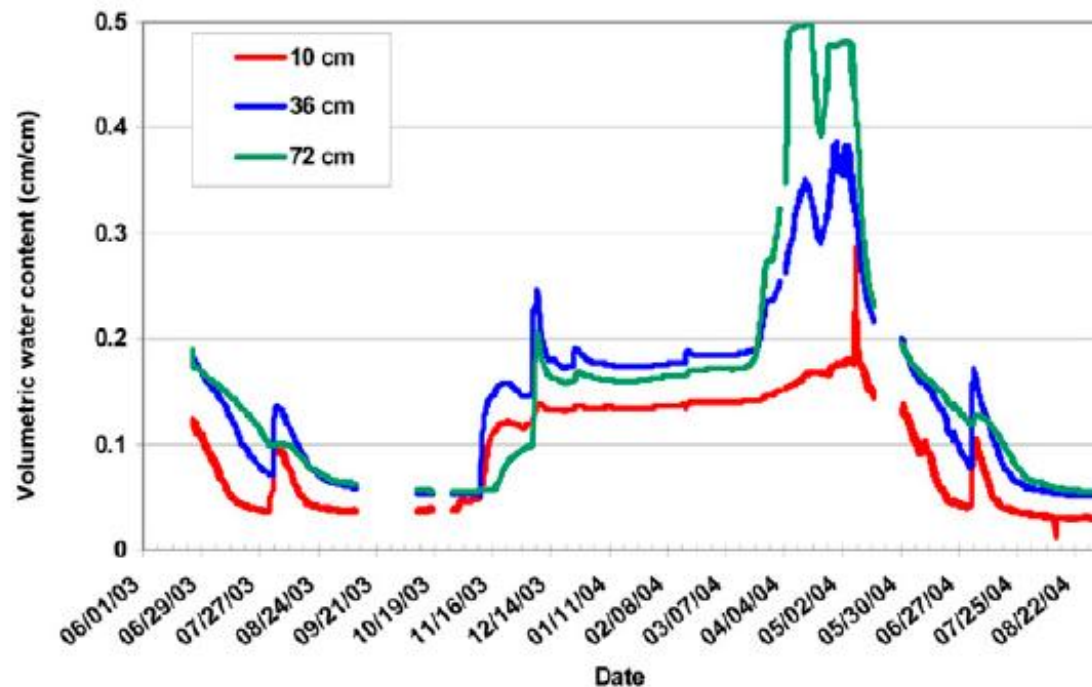
Singleton and Moran, WRR 2009

<sup>1</sup>Manning and Caine, 2007; <sup>2</sup>Cey et al., 2008

- RTs consistent with or slightly higher than MAATs
- Mean RT (7.8C) matches monthly mean air temperature for May (7.7C)
- Under current conditions, most recharge likely occurs during May-June







- Data from an instrumented soil zone at Gin Flat, Yosemite
- Rapid increase in soil temps and SWC once snowpack melts
- Nightly freeze allows soil to drain into weathered granite

# Findings: Recharge location and residence time

- Recharge occurs on lower slopes of catchment
  - Recharge temperatures close to mean annual air temperature and higher than expected for direct infiltration of snowmelt
  - Low excess air – minimal recharge through fractured rock
  - $\delta^{13}\text{C}$  of DIC indicates exchange with soil gas  $\text{CO}_2$
- Groundwater (even deep groundwater) in upstream portion of the basin is young

# Effects of Climate Change

## Climate Change Scenarios

- More precip as rain, extended period of runoff
- Earlier runoff
- More rain on snow events
- More nights above freezing temp.
- Less total precip

## Effect on Recharge and Discharge

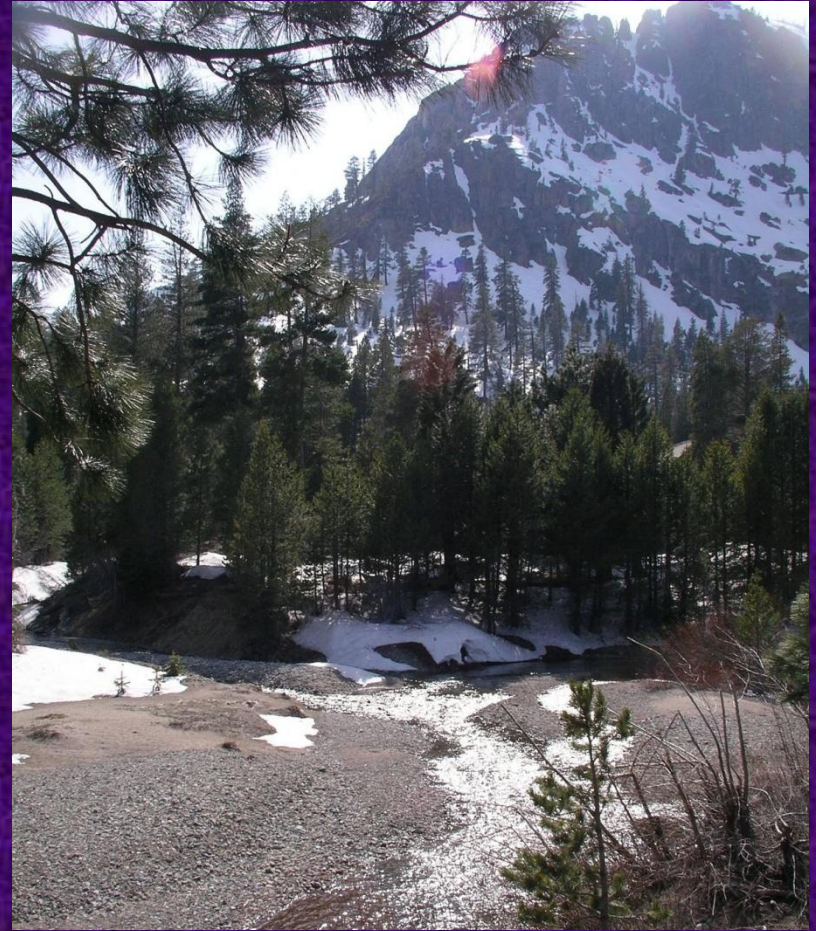
- More recharge, if precip rate is lower than current snowpack melt rate
- Early decreased baseflow (fast drainage)
- Increased overland flow, less recharge to alluvium
- More saturation-induced overland flow, less recharge
- Less recharge, near immediate effect on GW availability and streamflow

Effects will be immediate and drastic at Olympic Valley



# Acknowledgements

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  - Stable Isotope Lab (S. Roberts)
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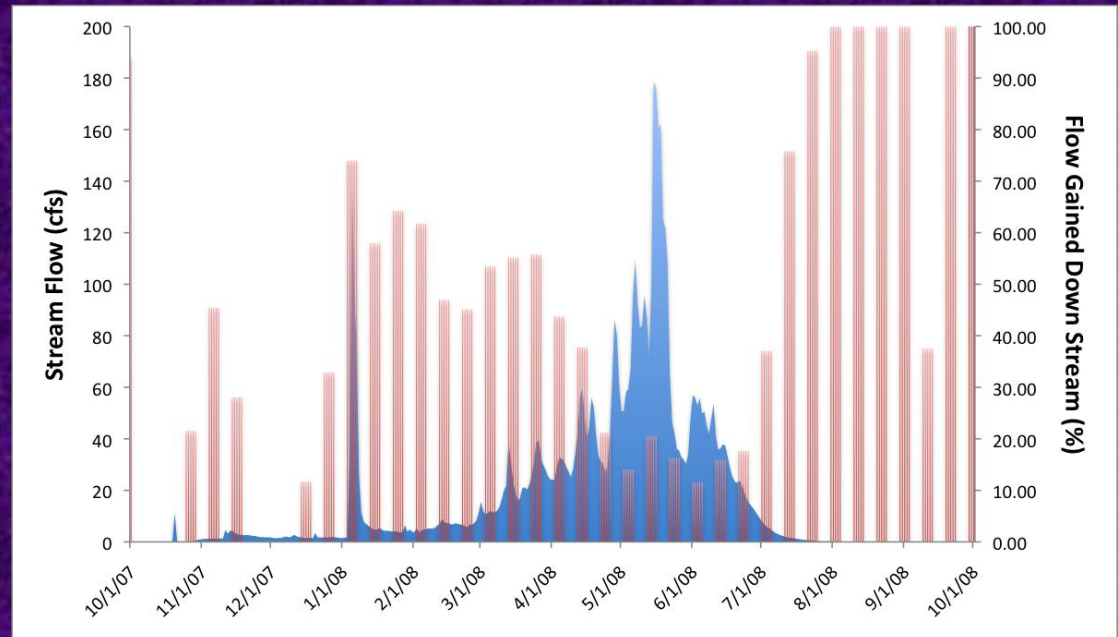
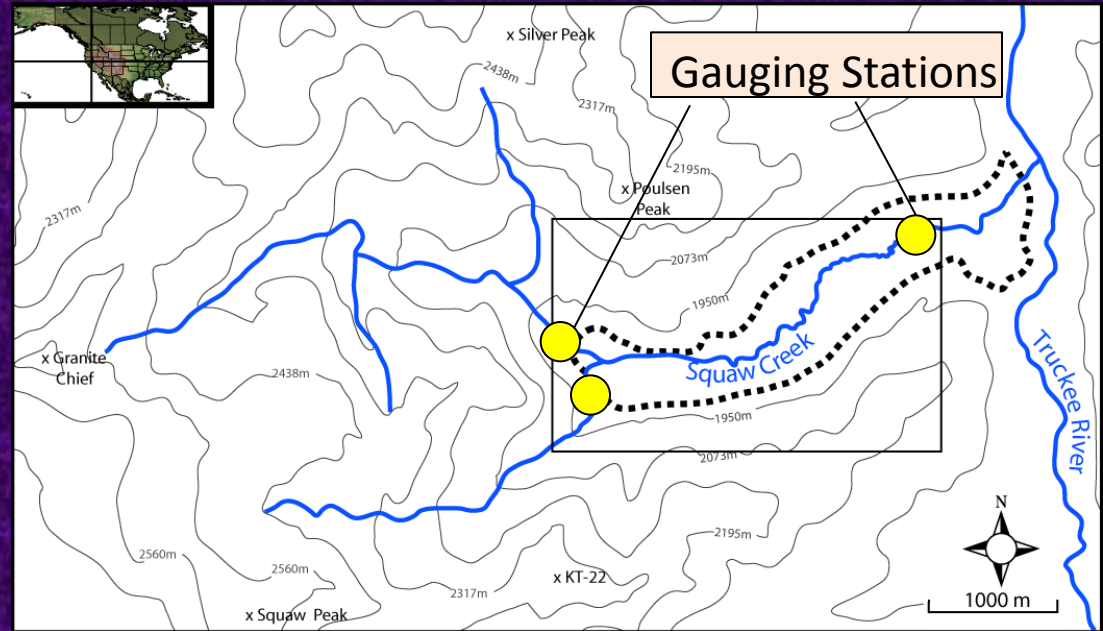


# Comparing three high elevation basins

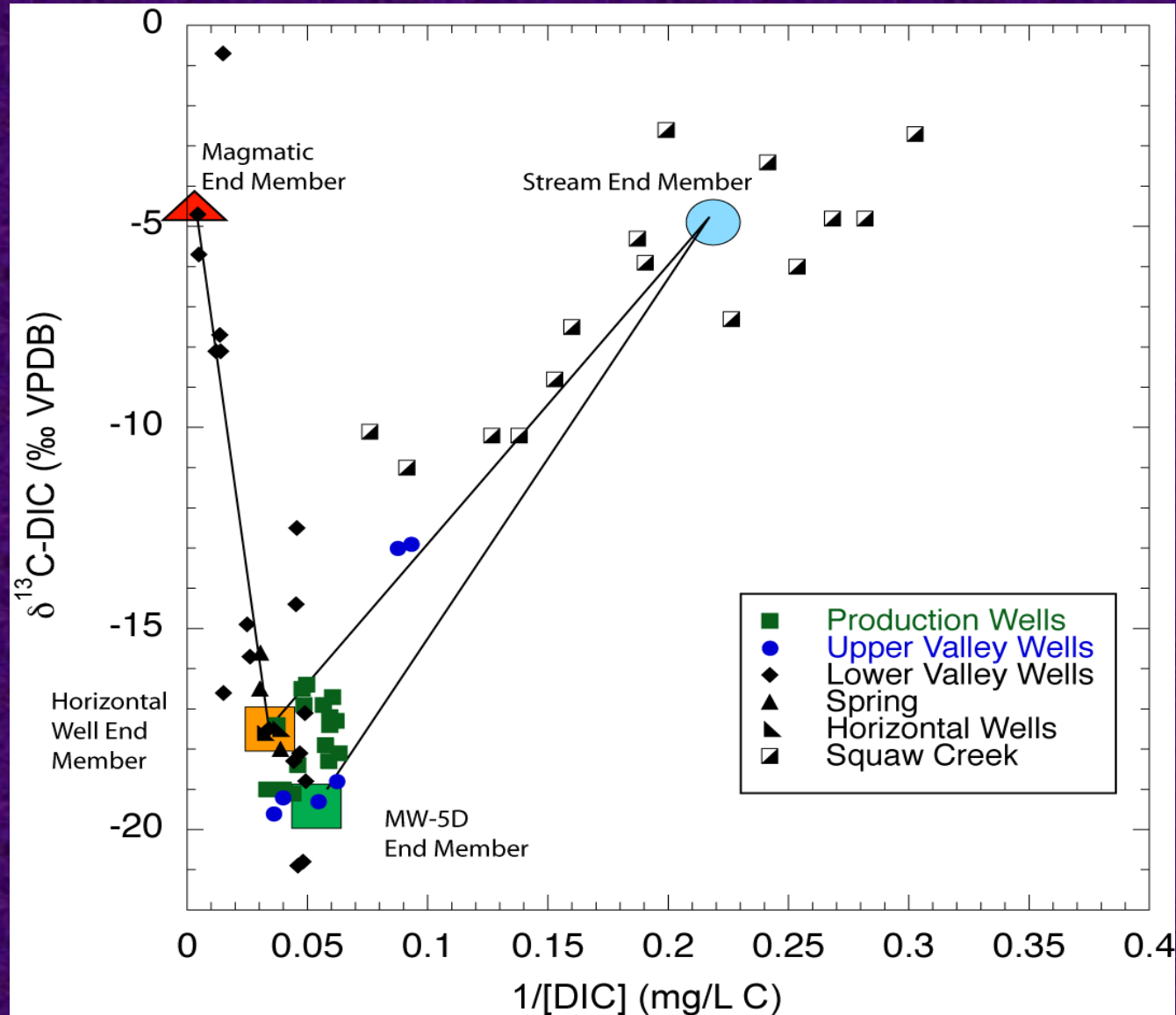
	Drainage Area	Max depth to bedrock	Valley floor elevation	Max elevation in drainage	Max discharge during study	Average annual discharge	Average Annual Precip
	GW Basin Area						
Olympic Valley	22 km <sup>2</sup>	55m	1898m – 1853m	2750m	5.2 m <sup>3</sup> /s	2x10 <sup>7</sup> m <sup>3</sup> /yr	1016-1650 mm
	2.6 km <sup>2</sup>						
Yosemite Valley	465 km <sup>2</sup>	600m (mean 300m)	1224m – 1100m (540m)	3997m	235 m <sup>3</sup> /s	64 x10 <sup>7</sup> m <sup>3</sup> /yr	1277 mm
	31 km <sup>2</sup>						
Martis Valley	433 km <sup>2</sup>	320 m	1737m-1798m	2624m	28 m <sup>3</sup> /s	50x10 <sup>7</sup> m <sup>3</sup> /yr	584-1910 mm
	142 km <sup>2</sup>						

# Squaw Creek

- Fed by two major tributaries
- Very low flow in the fall
- Gaining along the valley reach



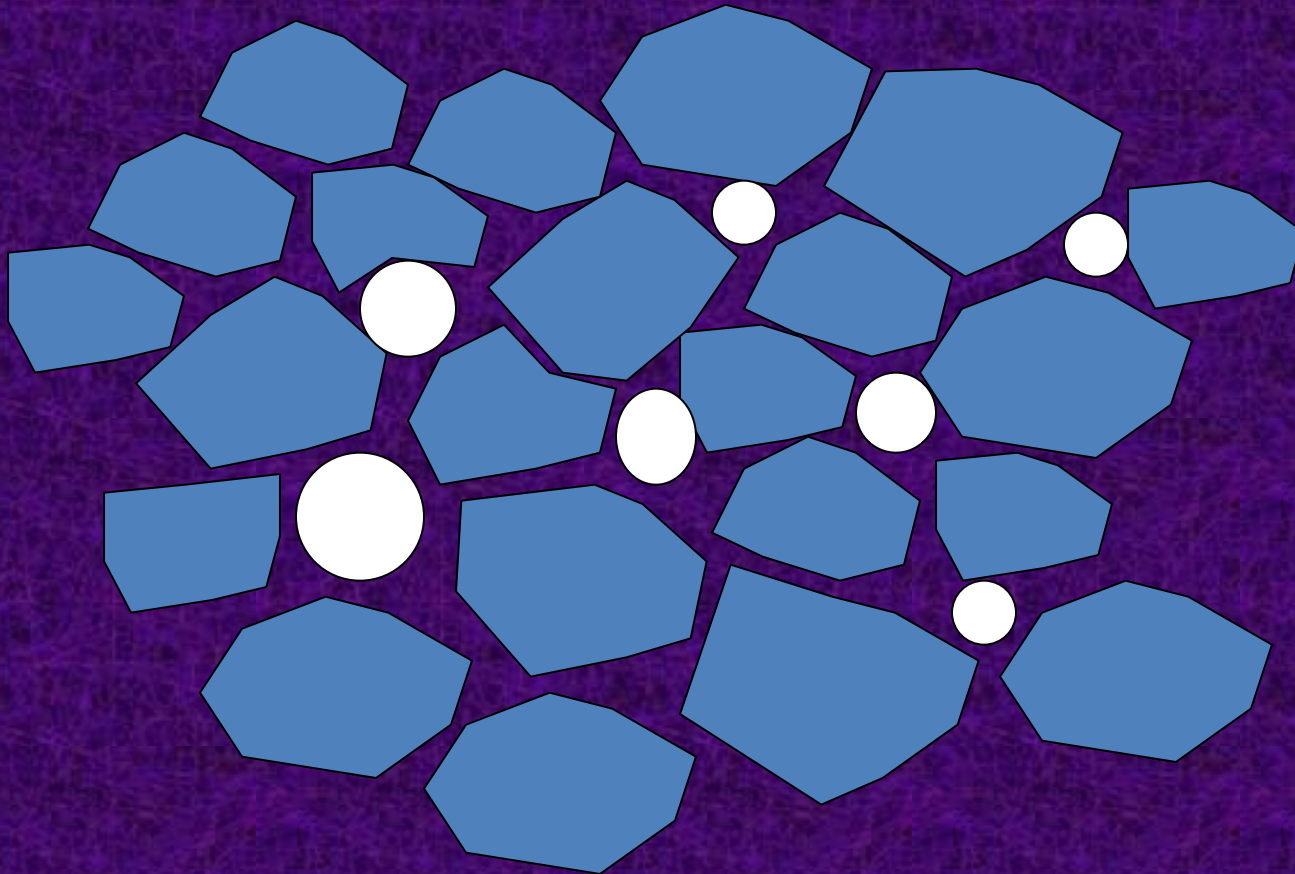
# Carbon isotopes are consistent with the incorporation of soil CO<sub>2</sub> during recharge



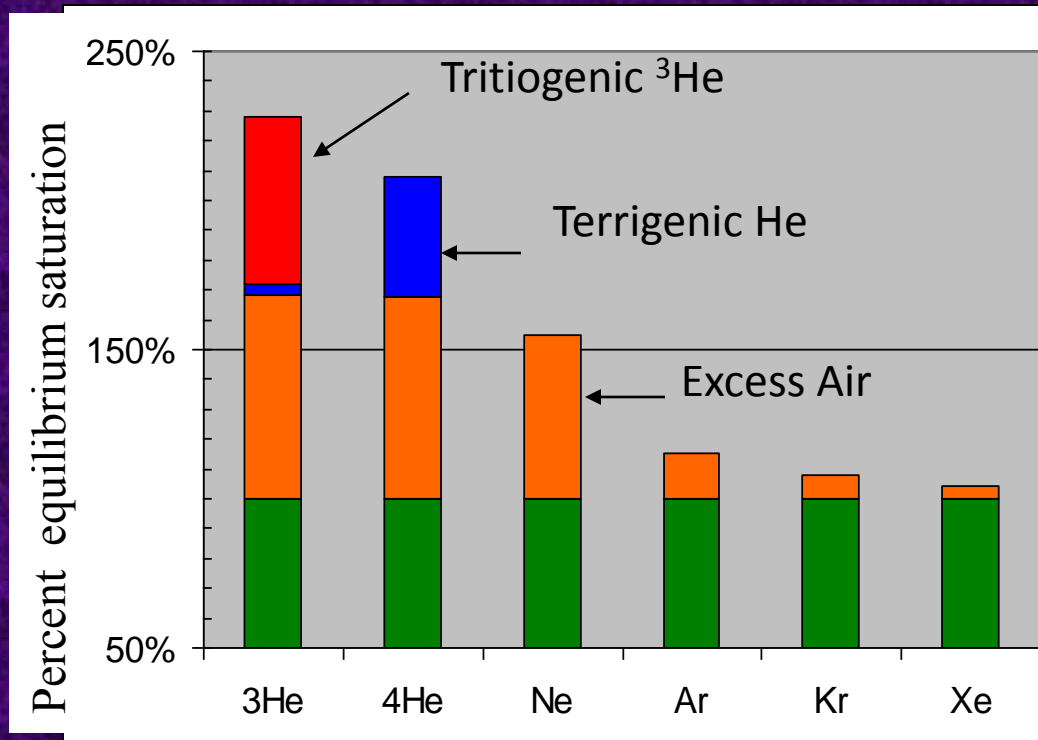


# What is excess air ?

Air bubbles can be trapped during recharge and subsequently dissolve because of the increased pressure



# Components of dissolved noble gases



- Equilibrium solubility (dependent upon  $p$ ,  $T$ )
- Excess Air
- Mantle Helium
- Terrigenous Helium
- Tritogenic helium (for  $^3\text{H}$ - $^3\text{He}$  age)

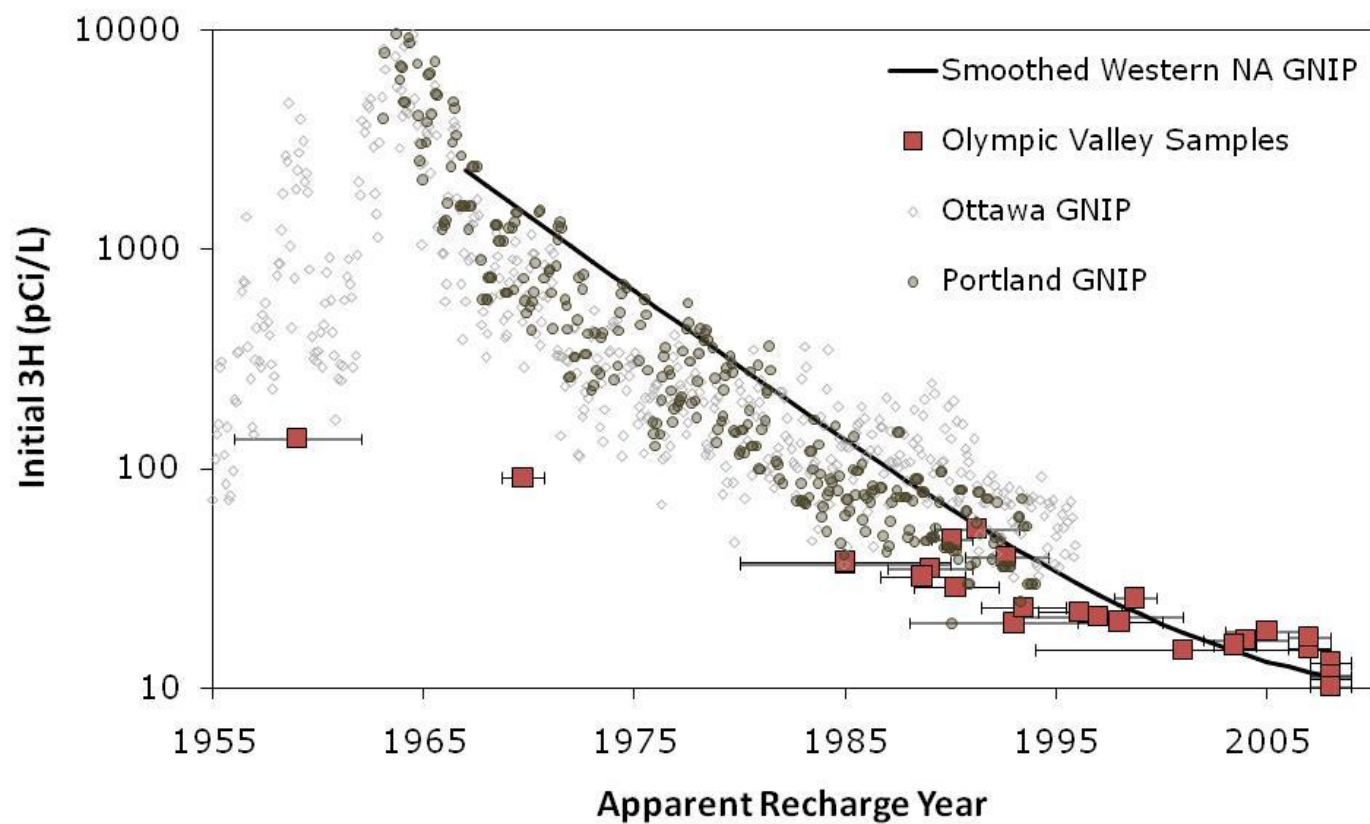
# Martis Valley sample locations



- 17 wells and 4 surface water locations sampled in Dec/Jan
- Most field work will take place this summer



# Fraction pre-modern



# Older groundwater is captured by wells during late summer

